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MEMORY ELEMENT FORMATION WITH

PHOTOSENSITIVE POLYMER DIELECTRIC

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MEMORY ELEMENT FORMATION WITH PHOTOSENSITIVE POLYMER DIELECTRIC

TECHNICAL FIELD

The present invention generally relates to organic memory elements containing a photosensitive dielectric and methods of making organic memory devices exploiting the properties of photosensitive dielectrics.

BACKGROUND ART

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The basic functions of a computer and memory devices include information processing and storage. In typical computer systems, these arithmetic, logic, and memory operations are performed by devices that are capable of reversibly switching between two states often referred to as "0" and "1." Such switching devices are fabricated from semiconducting devices that perform these various functions and are capable of switching between two states at high speed.

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Electronic addressing or logic devices, for instance for storage or processing of data, are made with inorganic solid state technology, and particularly crystalline silicon devices. The metal oxide semiconductor field effect transistor (MOSFET) is one the main workhorses.

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Much of the progress in making computers and memory devices faster, smaller and cheaper involves integration, squeezing ever more transistors and other electronic structures onto a postage-stamp-sized piece of silicon. A postage-stamp-sized piece of silicon may contain tens of millions of transistors, each transistor as small as a few hundred nanometers. However, silicon-based devices are approaching their fundamental physical size limits.

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Inorganic solid state devices are generally encumbered with a complex architecture which leads to high cost and a loss of data storage density. The circuitry of volatile semiconductor memories based on inorganic semiconductor material must constantly be supplied with electric current with a resulting heating and high electric power consumption in order to maintain stored information. Non-volatile

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semiconductor devices have a reduced data rate and relatively high power consumption and large degree of complexity.

Moreover, as inorganic solid state device sizes decrease and integration increases, sensitivity to alignment tolerances increases making fabrication markedly more difficult. Formation of features at small minimum sizes does not imply that the minimum size can be used for fabrication of working circuits. It is necessary to have alignment tolerances which are much smaller than the small minimum size, for example, one quarter the minimum size.

Scaling inorganic solid state devices raises issues with dopant diffusion lengths. As dimensions are reduced, the dopant diffusion lengths in silicon are posing difficulties in process design. In this connection, many accommodations are made to reduce dopant mobility and to reduce time at high temperatures. However, it is not clear that such accommodations can be continued indefinitely.

Applying a voltage across a semiconductor junction (in the reverse-bias direction) creates a depletion region around the junction. The width of the depletion region depends on the doping levels of the semiconductor. If the depletion region spreads to contact another depletion region, punch-through or uncontrolled current flow, may occur. Higher doping levels tend to minimize the separations required to prevent punch-through. However, if the voltage change per unit distance is large, further difficulties are created in that a large voltage change per unit distance implies that the magnitude of the electric field is large. An electron traversing such a sharp gradient may be accelerated to an energy level significantly higher than the minimum conduction band energy. Such an electron is known as a hot electron, and may be sufficiently energetic to pass through an insulator, leading to irreversibly degradation of a semiconductor device.

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Scaling and integration makes isolation in a monolithic semiconductor substrate more challenging. In particular, lateral isolation of devices from each other is difficult in some situations. Another difficulty is leakage current scaling. Yet another difficulty is presented by the diffusion of carriers within the substrate; that is free carriers can diffuse over many tens of microns and neutralize a stored charge.

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SUMMARY OF THE INVENTION

The following is a summary of the invention in order to provide a basic understanding of some aspects of the invention. This summary is not intended to identify key/critical elements of the invention or to delineate the scope of the invention. Its sole purpose is to present some concepts of the invention in a simplified form as a prelude to the more detailed description that is presented later.

The present invention provides organic memory devices that possess one or more of the following: simplified method of making the organic memory devices (mitigating the need for photoresist processing steps and etching steps), small size compared to inorganic memory devices, capability to store multiple bits of information, short resistance/impedance switch time, low operating voltages, low cost, high reliability, long life (thousands/millions of cycles), capable of three dimensional packing, associated low temperature processing, light weight, high density/integration, and extended memory retention.

One aspect of the present invention relates to an organic semiconductor memory device containing a patternable, photosensitive dielectric at least partially surrounding a plurality of organic semiconductor memory cells comprising a first electrode, a passive layer, an organic semiconductor layer, and a second electrode.

Another aspect of the present invention relates to a method of making an organic memory cell involving forming a photosensitive dielectric over a substrate, patterning the photosensitive dielectric to provide a patterned dielectric having openings therein, forming an organic semiconductor layer within the openings of the patterned dielectric.

To the accomplishment of the foregoing and related ends, the invention comprises the features hereinafter fully described and particularly pointed out in the claims. The following description and the annexed drawings set forth in detail certain illustrative aspects and implementations of the invention. These are indicative, however, of but a few of the various ways in which the principles of the invention may be employed. Other objects, advantages and novel features of the invention will

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become apparent from the following detailed description of the invention when considered in conjunction with the drawings.

BRIEF DESCRIPTION OF DRAWINGS

Figure 1 illustrates a perspective view of a two dimensional microelectronic device containing a plurality of organic memory cells in accordance with one aspect of the invention.

Figure 2 illustrates a cross-sectional view of a method of making organic memory cells in accordance with one aspect of the invention.

Figure 3 illustrates a cross-sectional view of a method of making organic memory cells in accordance with one aspect of the invention.

Figure 4 illustrates a cross-sectional view of a method of making organic memory cells in accordance with one aspect of the invention.

Figure 5 illustrates a cross-sectional view of a method of making organic memory cells in accordance with one aspect of the invention.

Figure 6 illustrates a cross-sectional view of a method of making organic memory cells in accordance with one aspect of the invention.

Figure 7 illustrates a perspective view of a three dimensional microelectronic device containing a plurality of organic memory cells in accordance with another aspect of the invention.

DISCLOSURE OF INVENTION

The present invention involves organic memory cells made of two electrodes with a controllably conductive media between the two electrodes. The controllably conductive media contains an organic semiconductor layer and passive layer. The organic semiconductor layer is made using spin-on techniques or chemical vapor deposition techniques, which permits the formation of inexpensive, efficient, and high quality organic semiconductor layers. The organic semiconductor layer is formed in vias/trenches formed using a photosensitive dielectric. The photosensitive dielectric can be patterned over electrodes without the use of a photoresist. Consequently, the

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need for CVD formed dielectric, dielectric etching steps, and/or photoresist processing steps are eliminated. Photoresist processing steps that are eliminated include photoresist deposition, photoresist stripping, application and removal of adhesion promotors, and the like. Cost and complexity of fabrication are thus improved.

The organic memory cells may optionally contain additional layers, such as additional electrodes, charge retention layers, and/or chemically active layers. The impedance of the controllably conductive media changes when an external stimuli such as an applied electric field is imposed. A plurality of organic memory cells, which may be referred to as an array, form an organic memory device. In this connection, organic memory cells may form an organic memory devices and function in a manner analogous to metal oxide semiconductor field effect transistors (MOSFETs) in conventional semiconductor memory devices. However, there are advantages to using the organic memory cells instead of conventional MOSFETs in memory devices.

Referring to Figure 1, a brief description of a microelectronic organic memory device 100 containing a plurality of organic memory cells in accordance with one aspect of the invention is shown, as well as an exploded view 102 of an exemplary organic memory cell 104. The microelectronic organic memory device 100 contains a desired number of organic memory cells, as determined by the number of rows, columns, and layers (three dimensional orientation described later) present. The first electrodes 106 and the second electrodes 108 are shown in substantially perpendicular orientation, although other orientations are possible to achieve the structure of the exploded view 102. Each organic memory cell 104 contains a first electrode 106 and a second electrode 108 with a controllably conductive media 110 therebetween. The controllably conductive media 110 contains an organic semiconductor layer 112 and passive layer 114. Peripheral circuitry and devices are not shown for brevity.

The photosensitive dielectric can be patterned without the use of a photoresist. That is, the photosensitive dielectric is deposited on the substrate (by spin-on, growth, or CVD techniques), selected portions are exposed to radiation, and a developer is employed to remove either the exposed or unexposed portions. General examples of

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photosensitive dielectrics include polyimides and photosensitive polyimides, polyimide-polyacrylates and photosensitive polyimide-polyacrylates, polyolefins and photosensitive polyolefins, fluorocarbon polymers and photosensitive fluorocarbon polymers, benzocyclobutene and photosensitive benzocyclobutene (BCB), organosilicons and photosensitive organosilicons, and hybrid organic/inorganic polymer and photosensitive hybrid organic/inorganic polymer. The photosensitive dielectric polymers may be photosensitive in and of themselves, or a photosensitive compound is intermixed with the polymer.

A polyimide having photosensitivity has a photosensitive functional group incorporated in the molecule of a polyimide precursor, and photoreaction occurs only in an exposed portion for polymerization, thereby varying the solubility between the unexposed portion and the exposed portion, and development is then conducted by making use of a solvent to dissolve and remove the unexposed portion while leaving only the exposed portion. If the photosensitive groups have poor heat resistance, the photosensitive groups are thermally decomposed and removed with the advance of cyclodehydration, so that only the polyimide portions having good heat resistance remain. The photosensitive groups can be introduced into the polyimide precursor through a covalent bond, an ionic bond or the like, and various photosensitive polyimides are commercially available. Polyimide precursors include precursors of polyimides, precursors of polyamide-imides, precursors of polybismaleimides, and precursors of modified polybismaleimides.

The photosensitive dielectrics include photosensitive polyimide-acrylates. The acrylic or methacrylic monomer or oligomer has at least two functionalities (in other words, at least two reactive groups having an unsaturated bond) in their molecules; namely, two of an isocyanurate structure, an acryloyl or methacryloyl group in a terminal and/or a side chain, oligoester acrylates and phosphazenic monomers or oligomers. In particular, acrylic or methacrylic monomers or oligomers which are commonly used as ultraviolet light curable adhesives are suitable.

For example, polyester acrylate, epoxy acrylate, urethane acrylate and silicone acrylate monomers or oligomers are useful as the acrylic or methacrylic monomer or

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oligomer. In particular, polyfunctional acrylate monomers or oligomers having an isocyanurate structure, for example, tris(acryloyloxyethyl) isocyanurate, isocyanuric acid EO (n=3) epsilon.-caprolactone modified triacrylate, and branched polyfunctional acrylate monomers or oligomers, for example, trimethylolpropane triacrylate, EO-modified trimethylolpropane triacrylate, pentaerythritol triacrylate, pentaerythritol tetraacrylate, and pentaerythritol hexaacrylate are useful. Further, methacrylic monomers or oligomers also can be similarly used. Acrylic or methacrylic fluoromonomers or fluorooligomers may be employed.

Specific examples include the residues formed by removing a hydrogen atom from a hydroxyl group in methacrylates such as 2-hydroxyethyl methacrylate, 2-hydroxypropyl methacrylate and 2-hydroxymethyl methacrylate (a hydroxyalkyl methacrylate residue), and residues formed by removing a hydrogen atom from a hydroxyl group in acrylates such as 2-hydroxyethyl acrylate, 2-hydroxybutyl acrylate and 3-hydroxy-2-tert-butylpropyl acrylate (a hydroxyalkyl acrylate residue).

When photosensitive polyimide-acrylates are employed, the photosensitive compound is preferably a polymerization initiator capable of generating radicals through the action of light, for example, benzoin ether compounds, ketalin ether compounds, ketal compounds, acetophenone compounds, benzophenone compounds, thioxanthone compounds, organic peroxides, N-phenylglycine, triazine compounds and allene-iron complexes, are suitable.

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Specific examples of photosensitive compounds preferably used with photosensitive polyimide-acrylates include isopropyl benzoin ether; isobutyl benzoin ether; 1-hydroxycyclohexyl phenyl ketone; benzyl dimethyl ketal; acetophenone; 2-hydroxy-2-methylpropiophenone; 4-isopropyl-2-hydroxy-2-methylpropiophenone; benzophenone; methyl o-benzoylbenzoate; 2-methylthioxanthone; 2-

- chlorothioxanthone; ketone peroxide; peroxy ketal; hydroperoxide; dialkyl peroxide; diacyl peroxide; peroxy ester; peroxy dicarbonate;
 - 2,4,6-tris(trichloromethyl)-1,3,5-triazine;
 - 2,4-bis(trichloromethyl)-6-phenyl-1,3,5-triazine; benzene-cyclopentadienyl iron (II)

hexafluorophosphate; pyrene-cyclopentadienyl iron (II) hexafluoroantimonate; and naphthalene-cyclopentadienyl iron (II) hexafluorophosphate.

When fluorocarbon polymers or organosilicon films are employed as the photosensitive dielectrics, either spin-on or CVD deposition techniques may be used. In CVD, there is no organic solvent used during resist deposition. CVD includes pulsed plasma enhanced chemical vapor deposition (PECVD) and pyrolytic CVD as well as continuous PECVD. Plasma polymerization is a common method for depositing fluorocarbon and organosilicon thin films. PECVD uses continuous radio frequency (rf) power to excite the precursor gases in order to deposit films within the glow discharge region.

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With pulsed PECVD and pyrolytic CVD methods, the as-deposited films tend to have fewer crosslinking sites than their continuous PECVD counterparts. As a result, irradiation resulting in the production of cross-linking groups can result in dramatic differences in chemical structure. Increasing structural differences between the as-deposited and irradiated films leads to higher lithographic contrast. In addition, growth rates for the pyrolytic processes are very high, which is a desirable feature for commercialization.

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An example of a fluorocarbon polymer is poly(CF₂) which is made by the polymerization of the reactive diradical difluorocarbene (:CF₂). Hexafluoropropylene oxide (HFPO) is used as a thermal source of the radical difluorocarbene (:CF₂). Examples of the organosilicon films include organosilanes and organosiloxanes. Examples of organosilanes include tetraethylorthosilicate, diethylsilane, tetramethylsilane and trieyoxysilane. The organosilicon film may be derived from hecxamethylcyclotrisiloxane, octamethylcyclotetrasiloxane or a mixture thereof.

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When fluorocarbon polymers or organosilicon films are employed as the photosensitive dielectrics, a supercritical fluid (SCF) may be used as a developer, in addition to the other developers listed herein. A SCF is defined as any fluid at a temperature that is greater than its critical temperature and at a pressure that is greater than its critical pressure. Examples of supercritical fluids include carbon dioxide; mixtures of carbon dioxide with at least one of butane, pentane, toluene, cyclohexane,

acetonitrile, and methanol; 2,3-dimethylbutane; ethanol; n-hexane; propane; propane/water mixtures; sulfur hexafluoride; propane; and ethane. When fluorocarbon polymers or organosilicon films are employed as the photosensitive dielectrics, a dry plasma etch may be used as a developer, in addition to the other developers listed herein.

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Specific examples of commercially available photosensitive dielectrics include photosensitive polyimides available under the trade designations ZFPI 5000, ZFPI 7000, ZKT140, and ZBT120 from Zeon Corporation and under the trade designation Photoneece from Toray Industries; photosensitive polyolefins available under the trade designation ZCOAT 1000 from Zeon Corporation; photosensitive hybrid organic/inorganic polymer under the trade designation CHEMAT-B from Chemat Technology, Inc.

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Photosensitive compounds may be chosen from a wide variety of compounds known to form radicals or an acid upon exposure to activating radiation.

Photosensitive compounds include photoacid generators and photosensitive polymerization initiators capable of generating radicals through the action of light. Examples of photosensitive polymerization initiators are listed above.

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In one general embodiment, photoacid generating compounds include aromatic substituted halohydrocarbons (such as 1,1-bis(p-chlorophenyl)-2,2,2-trichloroethane), halo-substituted sulfur containing compounds, haloheterocyclic compounds, onium salts (such as diaryl diazonium salts), azides, triazines, sulfonated esters, and sulfonated ketones.

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In one embodiment, the photosensitive compounds or photoacid generator is one or more unsubstituted and symmetrically or unsymmetrically substituted diaryliodonium, aryldiazonium, triarylselenonium, triarylsulfonium salts, a nitrobenzyl ester, an s-triazine derivative, an o-quinone diazide sulfonic acid ester including 2,1,4-diazonaphthoquinone sulfonic acid esters and the 2,1,5-diazonaphthoquinone sulfonic acid esters. In another embodiment, the photoacid generator is a sulfonated ester including sulfonyloxy ketones. Examples include

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benzoin tosylate, t-butylphenyl alpha-(p-toluenesulfonyloxy)-acetate, and t-butyl alpha-(p-toluenesulfonyloxy)-acetate.

In yet another embodiment, the photoacid generator is an onium salt. Examples of suitable onium salts are diaryldiazonium salts and onium salts of Group Va and B, Ia and B and I of the Periodic Table, for example, halonium salts, quaternary ammonium, phosphonium and arsonium salts, aromatic sulfonium salts, aryl sulfonium salts, alkyl sulfonium salts, and sulfoxonium salts or seleonium salts.

sulfonium hexafluoroantimonate, diaryliodonium metal halides, and certain non-ionic acid generators such as tris(sulfonate) of pyrogallol, and N-sulfonyloxynaphthalimides. In this embodiment, specific examples include triphenylsulfonium triflate and N-sulfonyloxynaphthalimide generators such as N-camphorsulfonyloxynaphthalimide and N-pentafluorobenzenesulfonyloxnaphthalimide.

In still yet another embodiment, the photoacid generator includes triaryl

In yet another embodiment, the photoacid generator includes halogenated non-ionic, photoacid generating compounds such as, for example, 1,1-bis [p-chlorophenyl]-2,2,2-trichloroethane (DDT); 1,1-bis [p-methoxyphenyl]-2,2,2-trichloroethane; 1,2,5,6,9,10-hexabromocyclododecane; 1,10-dibromodecane; 1,1-bis[p-chloropehnyl]-2,2-dichloroethane; 4,4'-dichloro-2-(trichloromethyl) benzhydrol (Kelthane); hexachlorodimethyl sulfone; 2-chloro-6-(trichloromethyl) pyridine; 0,0-diethyl-0-(3,5,6-trichloro-2-pyridyl)phosphorothionate; 1,2,3,4,5,6-hexachlorocyclohexane; N(1,1-bis [p-chlorophenyl]-2,2,2-trichloroethyl)acetamide; tris [2,3-dibromopropyl] isocyanurate; 2,2-bis [p-chlorophenyl]-1,1-dichloroethylene; tris [trichloromethyl] s-triazine; and their isomers, analogs, homologs, and residual compounds.

In yet another embodiment, the photoacid generator includes sulfonium salts. Examples of sulfonium salts include tris(4-dimethylaminophenyl)sulfonium ptoluenesulfonate; bis(4-tert-butoxyphenyl)(4-dimethylaminophenyl)sulfonium ptoluenesulfonate; (4-tert-butoxyphenyl)bis(4-dimethylaminophenyl)sulfonium ptoluenesulfonate; bis(4-tert-butoxycarbonylmethyloxyphenyl)(4-dimethylaminophenyl)

phenyl)sulfonium p-toluenesulfonate; (4-tert-butoxycarbonylmethyloxyphenyl)bis(4dimethylamino-phenyl)sulfonium p-toluenesulfonate; bis(4-(1ethoxyethyloxy)phenyl)(4-dimethylaminophenyl)-sulfonium p-toluenesulfonate: (4-(1-ethoxyethyloxy)phenyl)bis(4-dimethylaminophenyl)-sulfonium p-toluenesulfonate: bis(4-tetrahydropyranyloxyphenyl)(4-dimethylaminophenyl)-sulfonium p-5 toluenesulfonate; (4-tetrahydrofuranyloxyphenyl)bis(4-dimethylaminophenyl)sulfonium p-toluenesulfonate; tris(4-dimethylaminophenyl)sulfonium pethylbenzenesulfonate; bis(4-tert-butoxyphenyl)(4-dimethylaminophenyl)sulfonium pethylbenzenesulfonate; (4-tert-butoxyphenyl)bis(4-dimethylaminophenyl)sulfonium pethylbenzenesulfonate: tris(4-dimethylaminophenyl)sulfonium 2.4-dimethylbenzene-10 sulfonate; bis(4-tert-butoxyphenyl)(4-dimethylaminophenyl)sulfonium 2,4dimethylbenzenesulfonate; (4-tert-butoxyphenyl)bis(4dimethylaminophenyl)sulfonium 2,4-dimethylbenzenesulfonate: (4-tertbutoxyphenyl)bis(4-dimethylaminophenyl)sulfonium p-tert-butylbenzenesulfonate; bis(4-tert-butoxycarbonylmethyloxyphenyl)(4-dimethylamino-phenyl)sulfonium 2.5-15 dimethylbenzenesulfonate; (4-tert-butoxycarbonylmethyloxyphenyl)bis(4dimethylamino-phenyl)sulfonium n-octylbenzenesulfonate; tris(4dimethylaminophenyl)sulfonium p-tert-butylbenzene-sulfonate: bis(4-tertbutoxyphenyl)(4-dimethylaminophenyl)sulfonium p-tert-butylbenzenesulfonate; (4tert-butoxyphenyl)bis(4-dimethylaminophenyl)sulfonium benzenesulfonate; bis(4-tert-20 butoxycarbonylmethyloxyphenyl)(4-dimethylamino-phenyl)sulfonium benzenesulfonate; (4-tert-butoxycarbonylmethyloxyphenyl)bis(4-dimethylaminophenyl)sulfonium benzenesulfonate; tris(3-dimethylaminophenyl)sulfonium ptoluenesulfonate; bis(4-tert-butoxyphenyl)(3-dimethylaminophenyl)sulfonium ptoluenesulfonate; (4-tert-butoxyphenyl)bis(3-dimethylaminophenyl)sulfonium p-25 toluenesulfonate; bis(4-tert-butoxycarbonylmethyloxyphenyl)(3-dimethylaminophenyl)sulfonium p-toluenesulfonate; (4-tert-butoxycarbonylmethyloxyphenyl)bis(3dimethylamino-phenyl)sulfonium p-toluenesulfonate; bis(4-(1ethoxyethyloxy)phenyl)(3-dimethylaminophenyl)-sulfonium p-toluenesulfonate; (4-(1-ethoxyethyloxy)phenyl)bis(3-dimethylaminophenyl)-sulfonium p-toluenesulfonate;

bis(4-tetrahydropyranyloxyphenyl)(3-dimethylaminophenyl)-sulfonium ptoluenesulfonate; (4-tetrahydrofuranyloxyphenyl)bis(3-dimethylaminop henyl)sulfonium p-toluenesulfonate; tris(3-dimethylaminophenyl)sulfonium pethylbenzenesulfonate; bis(4-tert-butoxyphenyl)(3-dimethylaminophenyl)sulfonium 2.4-dimethylbenzenesulfonate; (4-tert-butoxycarbonylmethyloxyphenyl)bis(3-5 dimethylamino-phenyl)sulfonium n-octylbenzenesulfonate; bis(4-tertbutoxyphenyl)(3-dimethylaminophenyl)sulfonium benzenesulfonate; (4-tertbutoxyphenyl)bis(3-dimethylaminophenyl)sulfonium p-tert-butylbenzenesulfonate: bis(4-tert-butoxycarbonylmethyloxyphenyl)(3-dimethylamino-phenyl)sulfonium benzenesulfonate, (4-tert-butoxycarbonylmethyloxyphenyl)bis(3-dimethylamino-10 phenyl)sulfonium benzenesulfonate; tris(4-(2-picolyloxy)phenyl)sulfonium ptoluenesulfonate; bis(4-tert-butoxyphenyl)(4-(2-picolyloxy)phenyl)sulfonium ptoluenesulfonate; (4-tert-butoxyphenyl)bis(4-(2-picolyloxy)phenyl)sulfonium ptoluenesulfonate; bis(4-tert-butoxycarbonylmethyloxyphenyl)(4-(2picolyloxy)phenyl)sulfonium p-toluenesulfonate; (4-tert-15 butoxycarbonylmethyloxyphenyl)bis(4-(2-picolyloxy)phenyl)sulfonium ptoluenesulfonate; bis(4-(1-ethoxyethyloxy)phenyl)(4-(2-picolyloxy)phenyl)-sulfonium p-toluenesulfonate; (4-(1-ethoxyethyloxy)phenyl)bis(4-(2-picolyloxy)phenyl)sulfonium p-toluenesulfonate; bis(4-tetrahydropyranyloxyphenyl)(4-(2picolyloxy)phenyl)-sulfonium p-toluenesulfonate: (4-20 tetrahydrofuranyloxyphenyl)bis(4-(2-picolyloxy)phenyl)-sulfonium ptoluenesulfonate; tris(4-(2-picolyloxy)phenyl)sulfonium p-ethylbenzene-sulfonate; tris(4-(2-picolyloxy)phenyl)sulfonium 2,4-dimethylbenzene-sulfonate; bis(4-tertbutoxyphenyl)(4-(2-picolyloxy)phenyl)sulfonium 2,4-dimethylbenzenesulfonate; (4tert-butoxyphenyl)bis(4-(2-picolyloxy)phenyl)sulfonium 2.4-25 dimethylbenzenesulfonate; (4-tert-butoxyphenyl)bis(4-(2picolyloxy)phenyl)sulfonium p-tert-butylbenzenesulfonate: bis(4-tertbutoxycarbonylmethyloxyphenyl)(4-(2-picolyloxy)phenyl)sulfonium 2,5dimethylbenzenesulfonate; (4-tert-butoxycarbonylmethyloxyphenyl)bis(4-(2picolyloxy)phenyl)sulfonium n-octylbenzenesulfonate; tris(4-(2-

picolyloxy)phenyl)sulfonium p-tert-butylbenzene-sulfonate; bis(4-tertbutoxyphenyl)(4-(2-picolyloxy)phenyl)sulfonium benzenesulfonate; (4-tertbutoxyphenyl)bis(4-(2-picolyloxy)phenyl)sulfonium benzenesulfonate; bis(4-tertbutoxycarbonylmethyloxyphenyl)(4-(2-picolyloxy)phenyl)sulfonium benzenesulfonate; (4-tert-butoxycarbonylmethyloxyphenyl)bis(4-(2-5 picolyloxy)phenyl)sulfonium benzenesulfonate; tris(4-(4-picolyloxy)phenyl)sulfonium p-toluenesulfonate; bis(4-tert-butoxyphenyl)(4-(4-picolyloxy)phenyl)sulfonium ptoluenesulfonate; (4-tert-butoxyphenyl)bis(4-(4-picolyloxy)phenyl)sulfonium ptoluenesulfonate: bis(4-tert-butoxycarbonylmethyloxyphenyl)(4-(4-picolyloxy)phenyl)sulfonium p-toluenesulfonate; (4-tert-butoxycarbonylmethyloxyphenyl)bis(4-10 (4-picolyloxy)-phenyl)sulfonium p-toluenesulfonate; bis(4-(1ethoxyethyloxy)phenyl)(4-(4-picolyloxy)phenyl)-sulfonium p-toluenesulfonate; (4-(1ethoxyethyloxy)phenyl)bis(4-(4-picolyloxy)phenyl)-sulfonium p-toluenesulfonate; bis(4-tetrahydropyranyloxyphenyl)(4-(4-picolyloxy)phenyl)-sulfonium ptoluenesulfonate; (4-tetrahydrofuranyloxyphenyl)bis(4-(4-picolyloxy)phenyl)-15 sulfonium p-toluenesulfonate; tris(4-(4-picolyloxy)phenyl)sulfonium p-ethylbenzenesulfonate; tris(4-(4-picolyloxy)phenyl)sulfonium 2,4-dimethylbenzene-sulfonate; bis(4-tert-butoxyphenyl)(4-(4-picolyloxy)phenyl)sulfonium 2,4dimethylbenzenesulfonate; (4-tert-butoxyphenyl)bis(4-(4picolyloxy)phenyl)sulfonium 2,4-dimethylbenzenesulfonate; (4-tert-20 butoxycarbonylmethyloxyphenyl)bis(4-(4-picolyloxy)-phenyl)sulfonium 2.4dimethylbenzenesulfonate; bis(4-tert-butoxyphenyl)(4-(4picolyloxy)phenyl)sulfonium p-tert-butylbenzenesulfonate: (4-tertbutoxyphenyl)bis(4-(4-picolyloxy)phenyl)sulfonium p-tert-butylbenzenesulfonate; bis(4-tert-butoxycarbonylmethyloxyphenyl)(4-(4-picolyloxy)-phenyl)sulfonium 2.5-25 dimethylbenzenesulfonate; (4-tert-butoxycarbonylmethyloxyphenyl)bis(4-(4picolyloxy)-phenyl)sulfonium n-octylbenzenesulfonate; tris(4-(4picolyloxy)phenyl)sulfonium p-tert-butylbenzene-sulfonate; bis(4-tertbutoxyphenyl)(4-(4-picolyloxy)phenyl)sulfonium benzenesulfonate; (4-tertbutoxyphenyl)bis(4-(4-picolyloxy)phenyl)sulfonium p-tert-butylbenzenesulfonate;

bis(4-tert-butoxycarbonylmethyloxyphenyl)(4-(4-picolyloxy)-phenyl)sulfonium benzenesulfonate; (4-tert-butoxycarbonylmethyloxyphenyl)bis(4-(4-picolyloxy)phenyl)sulfonium benzenesulfonate: tris(3-(4-picolyloxy)phenyl)sulfonium ptoluenesulfonate; bis(4-tert-butoxyphenyl)(3-(4-picolyloxy)phenyl)sulfonium ptoluenesulfonate; (4-tert-butoxyphenyl)bis(3-(4-picolyloxy)phenyl)sulfonium p-5 toluenesulfonate: bis(4-tert-butoxycarbonylmethyloxyphenyl)(3-(4-picolyloxy)phenyl)sulfonium p-toluenesulfonate; (4-tert-butoxycarbonylmethyloxyphenyl)bis(3-(4-picolyloxy)-phenyl)sulfonium p-toluenesulfonate: bis(4-(1ethoxyethyloxy)phenyl)(3-(4-picolyloxy)phenyl)-sulfonium p-toluenesulfonate; (4-(1ethoxyethyloxy)phenyl)bis(3-(4-picolyloxy)phenyl)-sulfonium p-toluenesulfonate; 10 bis(4-tetrahydropyranyloxyphenyl)(3-(4-picolyloxy)phenyl)-sulfonium ptoluenesulfonate; (4-tetrahydrofuranyloxyphenyl)bis(3-(4-picolyloxy)phenyl)sulfonium p-toluenesulfonate; tris(3-(4-picolyloxy)phenyl)sulfonium p-ethylbenzenesulfonate; tris(3-(4-picolyloxy)phenyl)sulfonium 2,4-dimethylbenzene-sulfonate; bis(4-tert-butoxyphenyl)(3-(4-picolyloxy)phenyl)sulfonium 2,4-15 dimethylbenzenesulfonate; (4-tert-butoxyphenyl)bis(3-(4picolyloxy)phenyl)sulfonium 2,4-dimethylbenzenesulfonate; bis(4-tertbutoxycarbonylmethyloxyphenyl)(3-(4-picolyloxy)-phenyl)sulfonium 2.4dimethylbenzenesulfonate; (4-tert-butoxycarbonylmethyloxyphenyl)bis(3-(4picolyloxy)-phenyl)sulfonium 2,4-dimethylbenzenesulfonate; (4-tert-20 butoxycarbonylmethyloxyphenyl)bis(3-(4-picolyloxy)-phenyl)sulfonium noctylbenzenesulfonate; tris(3-(4-picolyloxy)phenyl)sulfonium benzenesulfonate; bis(4-tert-butoxyphenyl)(3-(4-picolyloxy)phenyl)sulfonium benzenesulfonate: (4-tertbutoxycarbonylmethyloxyphenyl)bis(3-(4-picolyloxy)phenyl)sulfonium benzenesulfonate; tris(3-(2-picolyloxy)phenyl)sulfonium p-toluenesulfonate: bis(4-25 tert-butoxyphenyl)(3-(2-picolyloxy)phenyl)sulfonium p-toluenesulfonate; (4-tertbutoxyphenyl)bis(3-(2-picolyloxy)phenyl)sulfonium p-toluenesulfonate; bis(4-tertbutoxycarbonylmethyloxyphenyl)(3-(2-picolyloxy)-phenyl)sulfonium ptoluenesulfonate; (4-tert-butoxycarbonylmethyloxyphenyl)bis(3-(2-picolyloxy)phenyl)sulfonium p-toluenesulfonate; bis(4-(1-ethoxyethyloxy)phenyl)(3-(2-

picolyloxy)phenyl)-sulfonium p-toluenesulfonate; (4-(1-ethoxyethyloxy)phenyl)bis(3-(2-picolyloxy)phenyl)-sulfonium p-toluenesulfonate: bis(4tetrahydropyranyloxyphenyl)(3-(2-picolyloxy)phenyl)-sulfonium p-toluenesulfonate; (4-tetrahydrofuranyloxyphenyl)bis(3-(2-picolyloxy)phenyl)-sulfonium ptoluenesulfonate; tris(3-(2-picolyloxy)phenyl)sulfonium p-ethylbenzene-sulfonate; 5 tris(3-(2-picolyloxy)phenyl)sulfonium 2.4-dimethylbenzene-sulfonate; bis(4-tertbutoxyphenyl)(3-(2-picolyloxy)phenyl)sulfonium 2,4-dimethylbenzenesulfonate; (4tert-butoxycarbonylmethyloxyphenyl)bis(3-(2-picolyloxy)-phenyl)sulfonium 2.4dimethylbenzenesulfonate; (3-tert-butoxyphenyl)bis(3-(2picolyloxy)phenyl)sulfonium p-tert-butylbenzenesulfonate; bis(4-tert-10 butoxycarbonylmethyloxyphenyl)(3-(2-picolyloxy)-phenyl)sulfonium 4-tertbutylbenzenesulfonate; (4-tert-butoxycarbonylmethyloxyphenyl)bis(3-(2-picolyloxy)phenyl)sulfonium n-octylbenzenesulfonate; tris(3-(2-picolyloxy)phenyl)sulfonium benzenesulfonate; bis(4-tert-butoxyphenyl)(3-(2-picolyloxy)phenyl)sulfonium benzenesulfonate; (4-tert-butoxycarbonylmethyloxyphenyl)bis(3-(2-picolyloxy)-15 phenyl)sulfonium benzenesulfonate; tris(pyridin-3-yl)sulfonium p-toluenesulfonate; bis(4-tert-butoxyphenyl)(pyridin-3-yl)sulfonium p-toluenesulfonate: (4-tertbutoxyphenyl)bis(pyridin-3-yl)sulfonium p-toluenesulfonate; bis(4-tertbutoxycarbonylmethyloxyphenyl)(pyridin-3-yl)-sulfonium p-toluenesulfonate; (4-tertbutoxycarbonylmethyloxyphenyl)bis(pyridin-3-yl)-sulfonium p-toluenesulfonate; 20 bis(4-(1-ethoxyethyloxy)phenyl)(pyridin-3-yl)sulfonium p-toluenesulfonate; (4-(1ethoxyethyloxy)phenyl)bis(pyridin-3-yl)sulfonium p-toluenesulfonate: bis(4tetrahydropyranyloxyphenyl)(pyridin-3-yl)sulfonium p-toluenesulfonate; (4tetrahydrofuranyloxyphenyl)bis(pyridin-3-yl)sulfonium p-toluenesulfonate; tris(pyridin-3-yl)sulfonium p-ethylbenzenesulfonate; tris(pyridin-3-yl)sulfonium 2.4-25 dimethylbenzenesulfonate; bis(4-tert-butoxyphenyl)(pyridin-3-yl)sulfonium 2,4dimethyl-benzenesulfonate; (4-tert-butoxycarbonylmethyloxyphenyl)bis(pyridin-3-yl)sulfonium 2,4-dimethylbenzenesulfonate; (4-tert-butoxyphenyl)bis(pyridin-3yl)sulfonium p-tert-butylbenzenesulfonate; (4-tertbutoxycarbonylmethyloxyphenyl)bis(pyridin-3-yl)-sulfonium n-

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octylbenzenesulfonate; tris(pyridin-3-yl)sulfonium benzenesulfonate; bis(4-tertbutoxyphenyl)(pyridin-3-yl)sulfonium benzenesulfonate; (4-tertbutoxycarbonylmethyloxyphenyl)bis(pyridin-3-yl)-sulfonium benzenesulfonate; tris(pyridin-2-yl)sulfonium p-toluenesulfonate; bis(4-tert-butoxyphenyl)(pyridin-2yl)sulfonium p-toluenesulfonate; (4-tert-butoxyphenyl)bis(pyridin-2-yl)sulfonium ptoluenesulfonate; bis(4-tert-butoxycarbonylmethyloxyphenyl)(pyridin-2-yl)-sulfonium p-toluenesulfonate: (4-tert-butoxycarbonylmethyloxyphenyl)bis(pyridin-2-yl)sulfonium p-toluenesulfonate; bis(4-(1-ethoxyethyloxy)phenyl)(pyridin-2yl)sulfonium p-toluenesulfonate; (4-(1-ethoxyethyloxy)phenyl)bis(pyridin-2yl)sulfonium p-toluenesulfonate; bis(4-tetrahydropyranyloxyphenyl)(pyridin-2yl)sulfonium p-toluenesulfonate; (4-tetrahydrofuranyloxyphenyl)bis(pyridin-2yl)sulfonium p-toluenesulfonate; tris(pyridin-2-yl)sulfonium p-ethylbenzenesulfonate; tris(pyridin-2-yl)sulfonium 2,4-dimethylbenzenesulfonate; bis(4-tertbutoxyphenyl)(pyridin-2-yl)sulfonium 2,4-dimethylbenzenesulfonate; (4-tertbutoxyphenyl)bis(pyridin-2-yl)sulfonium 2,4-dimethylbenzenesulfonate: bis(3-tertbutoxycarbonylmethyloxyphenyl)(pyridin-2-yl)-sulfonium 2.4dimethylbenzenesulfonate; (4-tert-butoxycarbonylmethyloxyphenyl)bis(pyridin-2-yl)sulfonium 2,4-dimethylbenzenesulfonate; (4-tert-butoxyphenyl)bis(pyridin-2yl)sulfonium p-tert-butylbenzenesulfonate; (4-tertbutoxycarbonylmethyloxyphenyl)bis(pyridin-2-yl)-sulfonium noctylbenzenesulfonate; and tris(pyridin-2-yl)sulfonium benzenesulfonate.

In still yet another embodiment, the photoacid generator includes alkylsulfonium salts. Examples of alkylsulfonium salts include cyclohexylmethyl(2-oxocyclohexyl) sulfonium trifluoromethanesulfonate; dicyclohexyl(2-oxocyclohexyl)sulfonium trifluoromethanesulfonate; cyclopentylmethyl(2-oxocyclohexyl)sulfonium trifluoromethanesulfonate; and cycloheptylmethyl(2-oxocyclopentyl)sulfonium trifluoromethanesulfonate.

Any suitable wavelength of radiation may be employed to expose portions of the photosensitive dielectric. For example, radiation having wavelengths from about 1 nm to about 550 nm may be employed, including X-rays and electron beams (e-

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beams). Specific examples of actinic radiation includes wavelengths of 11 nm, 13 nm, 157 nm, 193 nm, 248 nm, 253 nm, 302 nm, 313 nm, 334 nm, 365 nm, 405 nm, 435 nm, 546 nm, X-rays, and e-beams.

When removing either the exposed or unexposed portion of the photosensitive dielectric, a liquid or gas that solubilizes one of the exposed or unexposed portion (but not the other) is employed. Examples of such materials that can be employed to pattern or develop the photosensitive dielectric layer include glycol ether esters, glycol ethers, furans, and alkyl alcohols containing from about 4 to about 7 carbon atoms, alkanes, other organic solvents such as mesitylene or toluene, aqueous hydroxide solutions, aqueous acidic solutions, organic acidic solutions, organic basic solutions, and the like. Selection of a developer depends upon at least one of whether the exposed or unexposed portion is removed (positive or negative tone) and the identity of the photosensitive dielectric.

Referring to Figures 2 to 6, an exemplary method of making the organic memory elements/cells using a patternable, photosensitive dielectric is shown.

Specifically referring to Figure 2, a structure including a substrate 200 has a conductive metal structures 202 formed thereover/therein. The conductive metal structures 202 may be lines or via islands. The conductive metal structures 202 form one of the electrodes in the subsequently formed organic memory elements/cells. The conductive metal layer 202 contains copper, although any of the below mentioned electrode materials can be employed. A photosensitive dielectric 204 is formed over the substrate 200 and conductive metal structures 202. The thickness of the photosensitive dielectric 204 is approximately equal to the thickness of the subsequently formed organic semiconductor layer 212.

Referring to Figure 3, the photosensitive dielectric 204 is patterned to form a patterned dielectric 206 having openings 208 using lithographic techniques. The openings 208 can be, for example, trenches or vias. Specifically, portions of the photosensitive dielectric 204 are exposed to actinic radiation, causing a chemical change in the photosensitive dielectric 204 in the irradiated portions. The irradiated portions and the non-irradiated portions have different solubilities in certain solvents,

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fluids, gases, etchant, or SCF developers. A solvent, fluid, gas, etchant, or SCF developer is contacted with the irradiated photosensitive dielectric 204 to remove either the irradiated portions or the non-irradiated portions to provide patterned dielectric 206 having openings 208. Consequently, the use of a photoresist is not necessary to pattern the photosensitive dielectric 204. The patterned dielectric 206 provides electrical insulation between subsequently formed organic memory cells.

In an alternative embodiment, although not shown, the photosensitive dielectric 204 is patterned to form a patterned dielectric 206 having openings 208, portions of the substrate 200 removed within the openings 208, a metal deposited over the substrate 200 and within the openings 208 to provide the same structure.

Referring to Figure 4, in the embodiment of Figure 3 or the alternative embodiment, the conductive metal structures 202 may be oxidized or otherwise treated to form a thin passive layer 210 containing, for example, copper sulfide, over the upper portion of the conductive metal structures 202.

Referring to Figure 5, an organic semiconductor layer 212 is formed over the substrate 200 and in particular within the opening 208 of the patterned dielectric 206. The organic semiconductor layer 212 contains an organic semiconductor material (such as a conductive polymer). The organic semiconductor layer 212 may be formed over the entire substrate 200, followed by a subtractive process to position the organic semiconductor layer 212 within the openings 208.

Referring to Figure 6, formation of the organic memory cells is completed by forming conductive metal structures 214 over the organic semiconductor structures 212. This can be done in a variety of ways, including forming a conductive metal layer over the structure, patterning the metal layer to correspond with the organic semiconductor structures 212, and forming additional dielectric material 216 over the structure, providing openings in the additional dielectric material 216 over the organic semiconductor structures 212, forming a conductive metal layer 214 in the opening, and adding further dielectric material to the structure over the conductive metal layer 214.

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The organic memory cells contain at least two electrodes, as one or more electrodes may be disposed between the two electrodes that sandwich the controllably conductive media. The electrodes are made of conductive material, such as conductive metal, conductive metal alloys, conductive metal oxides, conductive polymer films, semiconductive materials, and the like.

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Examples of electrodes include one or more of aluminum, chromium, copper, germanium, gold, magnesium, manganese, indium, iron, nickel, palladium, platinum, silver, titanium, zinc, and alloys thereof; indium-tin oxide (ITO); polysilicon; doped amorphous silicon; metal silicides; and the like. Alloy electrodes specifically include Hastelloy®, Kovar®, Invar, Monel®, Inconel®, brass, stainless steel, magnesium-silver alloy, and various other alloys.

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In one embodiment, the thickness of each electrode is independently about 0.01 μm or more and about 10 μm or less. In another embodiment, the thickness of each electrode is independently about 0.05 μm or more and about 5 μm or less. In yet another embodiment, the thickness of each electrode is independently about 0.1 μm or more and about 1 μm or less.

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The controllably conductive media, disposed between the two electrodes, can be rendered conductive or non-conductive in a controllable manner using an external stimuli. Generally, in the absence of an external stimuli, the controllably conductive media is non-conductive or has a high impedance. Further, in some embodiments, multiple degrees of conductivity/resistivity may be established for the controllably conductive media in a controllable manner. For example, the multiple degrees of conductivity/resistivity for the controllably conductive media may include a non-conductive state, a highly conductive state, and a semiconductive state.

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The controllably conductive media can be rendered conductive, nonconductive or any state therebetween (degree of conductivity) in a controllable manner by an external stimulus (external meaning originating from outside the controllably conductive media). For example, under an external electric field, radiation, and the like, a given non-conductive controllably conductive media is converted to a conductive controllably conductive media.

The controllably conductive media contains one or more organic semiconductor layers and one or more passive layers. In one embodiment, the controllably conductive media contains at least one organic semiconductor layer that is adjacent a passive layer (without any intermediary layers between the organic semiconductor layer and passive layer).

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The organic semiconductor layer contains an organic semiconductor material selected from at least one of an organic polymer (such as a conjugated organic polymer), an organometallic polymer (such as a conjugated organometallic polymer), a buckyball, a carbon nanotube (such as a C6-C60 carbon nanotubes), and the like. Organic semiconductors thus have a carbon based structure, often a carbon-hydrogen based structure, which is different from conventional MOSFETs. The organic semiconductor materials are typically characterized in that they have overlapping π orbitals, and/or in that they have at least two stable oxidation states. The organic semiconductor materials are also characterized in that they may assume two or more resonant structures. The overlapping π orbitals contribute to the controllably conductive properties of the controllably conductive media. The amount of charge injected into the organic semiconductor layer also influences the degree of conductivity of the organic semiconductor layer.

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A carbon nanotube is typically a hexagonal network of carbon atoms (from about 6 to about 60 carbon atoms, typically) that is rolled up into a seamless cylinder. Each end may be capped with half of a fullerene molecule. Carbon nanotubes may be prepared by the laser vaporization of a carbon target (a cobalt-nickel catalyst may facilitate growth) or a carbon-arc method to grow similar arrays of single-wall nanotubes. A buckyball is more specifically a buckminster-fullerene, a soccerball-shaped 60-atom cluster of pure carbon.

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The organic polymer typically contains a conjugated organic polymer. The polymer backbone of the conjugated organic polymer extends lengthwise between the electrodes (generally substantially perpendicular to the inner, facing surfaces of the electrodes). The conjugated organic polymer may be linear or branched, so long as the polymer retains its conjugated nature. Conjugated polymers are characterized in

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that they have overlapping π orbitals. Conjugated polymers are also characterized in that they may assume two or more resonant structures. The conjugated nature of the conjugated organic polymer contributes to the controllably conductive properties of the controllably conductive media.

In this connection, the organic semiconductor layer, such as the conjugated organic polymer, has the ability to donate and accept charges (holes and/or electrons). Generally, the organic semiconductor or an atom/moiety in the polymer has at least two relatively stable oxidation states. The two relatively stable oxidation states permit the organic semiconductor to donate and accept charges and electrically interact with the conductivity facilitating compound. The ability of the organic semiconductor layer to donate and accept charges and electrically interact with the passive layer also depends on the identity of the conductivity facilitating compound.

The organic polymers (or the organic monomers constituting the organic polymers) may be cyclic or acyclic. Examples of conjugated organic polymers include one or more of polyacetylene; polyphenylacetylene; polydiphenylacetylene; polyaniline; poly(p-phenylene vinylene); polythiophene; polyporphyrins; porphyrinic macrocycles, thiol derivatized polyporphyrins; polymetallocenes such as polyferrocenes, polyphthalocyanines; polyvinylenes; polystiroles; and the like.

Chemical structures of examples of repeating units/moieties that make up the conjugated organic polymers and conjugated organometallic polymers include one or more of Formulae (I) to (XIII):

$$\begin{array}{c|c}
\hline
\end{array}$$
(I)

$$\begin{array}{c|c} & & & \\ \hline & & \\ \hline & & \\ \hline & & \\ \hline & & \\ \end{array}$$

$$S$$
 n
 (IV)

$$\begin{array}{c|c} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\$$

wherein each R is independently hydrogen or hydrocarbyl; each M is independently a metal; each E is independently O, N, S, Se, Te, or CH; each L is independently a group containing or continuing conjugation (unsaturation); and each n is independently about 1 or more and about 25,000 or less. In another embodiment, each n is independently about 2 or more and about 10,000 or less. In yet another embodiment, each n is independently about 20 or more and about 5,000 or less. Examples of metals include Ag, Al, Au, B, Cd, Co, Cu, Fe, Ga, Hg, Ir, Mg, Mn, Ni, Pb, Pd, Pt, Rh, Sn, and Zn. Examples of L groups include hydrocarbyl groups possessing conjugation or the ability to form resonance structures, such as phenyl groups, substituted phenyl groups, acetylene groups, and the like.

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Any of the formulae may have one or more pendent substituent groups, not shown in the formulae. For example, a phenyl group may appear on the polythiophene structure, such as on the 3 position of each thiophene moiety. As another example, alkyl, alkoxy, cyano, amino, and/or hydroxy substituent groups may be present on the phenyl rings in any of the polyphenylacetylene, polydiphenylacetylene, and poly(p-phenylene vinylene) conjugated polymers. Alternatively, any of the formulae may have one or more acid labile pendent substituent groups (described further in detail below).

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The term "hydrocarbyl" includes hydrocarbon as well as substantially hydrocarbon groups. Hydrocarbyl groups contain 1 or more carbon atom and typically about 60 or less carbon atoms. In another embodiment, hydrocarbyl groups contain 2 or more carbon atoms and about 30 or less carbon atoms. Substantially hydrocarbon describes groups which contain heteroatom substituents or heteroatoms which do not alter the predominantly organic character of the polymer, and do not impede the ability of the organic polymer to form a conjugated structure. Examples of hydrocarbyl groups include the following:

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(1) hydrocarbon substituents, i.e., aliphatic (e.g., alkyl or alkenyl), alicyclic (e.g., cycloalkyl, cycloalkenyl) substituents, acyl, phenyl, aromatic-, aliphatic- and alicyclic-substituted aromatic substituents and the like as well as cyclic substituents

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wherein the ring is completed through another portion of the molecule (that is, for example, any two indicated substituents may together form an alicyclic radical);

- (2) substituted hydrocarbon substituents, i.e., those substituents containing non-hydrocarbon groups which, in the context of this invention, do not alter the predominantly organic nature of the substituent; those skilled in the art will be aware of such groups (e.g., halo (especially chloro and fluoro, such as perfluoroalkyl, perfluoroaryl), cyano, thiocyanato, amino, alkylamino, sulfonyl, hydroxy, mercapto, nitro, nitroso, sulfoxy, etc.);
- (3) heteroatom substituents, i.e., substituents which, while having a predominantly organic character within the context of this invention, contain an atom other than carbon present in a ring or chain otherwise composed of carbon atoms (e.g., alkoxy, alkylthio). Suitable heteroatoms will be apparent to those of ordinary skill in the art and include, for example, sulfur, oxygen, nitrogen, fluorine, chlorine, and such substituents as, e.g., pyridyl, furyl, thienyl, imidazolyl, imido, amido, carbamoyl, etc.

In one embodiment, the organic semiconductor layer contains a thin layer designed to improve or lengthen charge retention time. The thin layer may be disposed anywhere within the organic semiconductor layer, but typically near the middle of the layer. The thin layer contains any of the electrode materials or the compounds of the below-described heterocyclic/aromatic compound layer. In one embodiment, the thin layer has a thickness of about 50 Å or more and about 0.1 µm or less. In another embodiment, the thin layer has a thickness of about 100 Å or more and about 0.05 µm or less. For example, an organic memory cell may contain a first electrode of copper, a passive layer of copper sulfide, an organic semiconductor layer of poly(phenylene vinylene), and a second electrode of aluminum, wherein the poly(phenylene vinylene) organic semiconductor layer contains a 250 Å thick layer of copper therein.

In one embodiment, the organic semiconductor material does not contain an organometallic compound. In another embodiment, the organic semiconductor material contains an organic polymer doped with an organometallic compound.

In one embodiment, the organic semiconductor layer has a thickness of about 0.001 μ m or more and about 5 μ m or less. In another embodiment, the organic semiconductor layer has a thickness of about 0.01 μ m or more and about 2.5 μ m or less. In yet another embodiment, the organic semiconductor layer has a thickness of about 0.05 μ m or more and about 1 μ m or less.

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The organic semiconductor layer may be formed by spin-on techniques (depositing a mixture of the polymer/polymer precursor and a solvent, then removing the solvent from the substrate/electrode) or CVD techniques. During formation, the organic semiconductor material self assembles between the electrodes. It is not typically necessary to functionalize one or more ends of the organic polymer in order to attach it to an electrode/passive layer. When using spin-on techniques, the solvent in which the polymer/polymer precursor is charged before application to the wafer structure is able to solubilize the polymer/polymer precursor so that the organic semiconductor layer is formed in a substantially uniform manner over the surface of the substrate.

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Solvents useful for this purpose include one or more of glycol ether esters, glycol ethers, furans, and alkyl alcohols containing from about 4 to about 7 carbon atoms. Two or more of glycol ether esters, glycol ethers, furans, and alkyl alcohols containing from about 4 to about 7 carbon atoms may be employed as a solvent system. The solvent system may contain one or more of glycol ether esters, glycol ethers, furans, and alkyl alcohols, and another organic solvent.

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Examples of glycol ether esters include ethylene glycol methyl ether acetate, ethylene glycol ethyl ether acetate, ethylene glycol propyl ether acetate, ethylene glycol butyl ether acetate, propylene glycol methyl ether acetate, propylene glycol ethyl ether acetate, propylene glycol propyl ether acetate, propylene glycol butyl ether acetate, and the like.

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Glycol ether esters also include polyalkylene glycol ether esters such as poly(ethylene glycol) alkyl ether acetates and poly(propylene glycol) alkyl ether acetates. Examples of poly(ethylene glycol) alkyl ether acetates and poly(propylene glycol) alkyl ether acetates include poly(ethylene glycol) methyl ether acetate,

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poly(ethylene glycol) ethyl ether acetate, poly(ethylene glycol) propyl ether acetate, poly(ethylene glycol) butyl ether acetate, poly(propylene glycol) methyl ether acetate, poly(propylene glycol) ethyl ether acetate, poly(propylene glycol) propyl ether acetate, and poly(propylene glycol) butyl ether acetate.

Additional specific examples of polyalkylene glycol ether esters include di(ethylene glycol) methyl ether acetate, di(ethylene glycol) ethyl ether acetate, di(ethylene glycol) butyl ether acetate, di(ethylene glycol) butyl ether acetate, di(ethylene glycol) dodecyl ether acetate, di(propylene glycol) methyl ether acetate, di(propylene glycol) ethyl ether acetate, di(propylene glycol) butyl ether acetate, tri(ethylene glycol) methyl ether acetate, tri(ethylene glycol) butyl ether acetate, tri(propylene glycol) methyl ether acetate, tri(propylene glycol) methyl ether acetate, acetate, tri(propylene glycol) butyl ether acetate, and the like.

Examples of glycol ethers include alkylene glycol ethers and polyalkylene glycol ethers, such as poly(ethylene glycol) methyl ether, poly(ethylene glycol) ethyl ether, poly(ethylene glycol) propyl ether, poly(ethylene glycol) butyl ether, poly(propylene glycol) methyl ether, poly(propylene glycol) ethyl ether, poly(propylene glycol) propyl ether, and poly(propylene glycol) butyl ether. Other examples of glycol ethers include ethylene glycol methyl ether, ethylene glycol methylbutyl ether, ethylene glycol ethylbutyl ether, ethylene glycol ethyl ether, ethylene glycol butyl ether, ethylene glycol dimethyl ether, ethylene glycol diethyl ether, ethylene glycol dibutyl ether, propylene glycol methyl ether, propylene glycol ethyl ether, propylene glycol butyl ether, propylene glycol dimethyl ether, propylene glycol diethyl ether, propylene glycol dibutyl ether, di(ethylene glycol) methyl ether, di(ethylene glycol) ethyl ether, di(ethylene glycol) butyl ether, di(ethylene glycol) hexyl ether, di(ethylene glycol) dimethyl ether, di(ethylene glycol) diethyl ether, di(ethylene glycol) dibutyl ether, di(ethylene glycol) butylmethyl ether, di(ethylene glycol) dodecyl ether, di(propylene glycol) methyl ether, di(propylene glycol) butyl ether, tri(ethylene glycol) methyl ether, tri(ethylene glycol) dimethyl ether, tri(propylene glycol) methyl ether, and tri(propylene glycol) butyl ether.

Furans include tetrahydrofuran. Alkyl alcohols containing from about 4 to about 7 carbon atoms secifically include alkyl alcohols containing from about 5 to about 6 carbon atoms. Examples of alkyl alcohols containing from about 4 to about 7 carbon atoms include n-butanol, iso-butanol, n-pentanol, iso-pentanol, cyclopentanol, n-hexanol, cyclohexanol, heptanol, and the like.

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Using one or more of glycol ether esters, glycol ethers, furans, and alkyl alcohols containing from about 4 to about 7 carbon atoms, formation of the organic semiconductor layer is facilitated. In particular, forming the organic polymer in a direction away from the surface on which it is formed is facilitated, delivering the organic semiconductor material to the substrate surface is facilitated, uniformly spaced organic polymer backbones are formed on the surface, and/or the formation of the organic semiconductor layer proceeds in a controllable manner.

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In one embodiment, the mixture contains from about 0.1% to about 75% by weight of the organic semiconductor material and from about 25% to about 99.9% by weight of solvent (one or more of glycol ether esters, glycol ethers, furans, and alkyl alcohols). In another embodiment, the mixture contains from about 0.5% to about 50% by weight of the organic semiconductor material from about 50% to about 99.5% by weight of solvent. In yet embodiment, the mixture contains from about 1% to about 25% by weight of the organic semiconductor material and from about 75% to about 99% by weight of solvent.

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When applying the mixture of organic semiconductor material and solvent to a wafer structure, the mixture is at a temperature suitable to facilitate formation of the organic semiconductor layer, facilitate solubilization of the organic semiconductor material and/or permit easy removal of the solvent from the wafer substrate. In one embodiment, the temperature of the mixture during application is from about 15 °C. to about 80° C. In another embodiment, the temperature of the mixture during application is from about 25 °C. to about 70° C. In yet another embodiment, the temperature of the mixture during application is from about 30 °C. to about 60° C.

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A covalent bond may be formed between the organic semiconductor material and the passive layer. Alternatively, close contact is required to provide good charge

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carrier/electron exchange between the organic semiconductor layer and the passive layer. The organic semiconductor layer and the passive layer are electrically coupled in that charge carrier/electron exchange occurs between the two layers.

A passive layer contains at least one conductivity facilitating compound that contributes to the controllably conductive properties of the controllably conductive media. The conductivity facilitating compound has the ability to donate and accept charges (holes and/or electrons). The passive layer thus may transport between an electrode and the organic polymer layer/passive layer interface, facilitate charge/carrier injection into the organic polymer layer, and/or increase the concentration of a charge carrier in the organic polymer layer. In some instances, the passive layer may store opposite charges thereby providing a balance of charges in the organic memory device as a whole. Storing charges/charge carriers is facilitated by the existence of two relatively stable oxidation states for the conductivity facilitating compound.

Generally, the conductivity facilitating compound or an atom in the conductivity facilitating compound has at least two relatively stable oxidation states. The two relatively stable oxidation states permit the conductivity facilitating compound to donate and accept charges and electrically interact with the organic semiconductor layer. The particular conductivity facilitating compound employed in a given organic memory cell is selected so that the two relatively stable oxidation states match with the two relatively stable oxidation states of the organic semiconductor material. Matching the energy bands of two relatively stable oxidation states of the organic semiconductor material and the conductivity facilitating compound facilitate charge carrier retention in the organic semiconductor layer.

Matching energy bands means that the fermi level of the passive layer is close to the valence band of the organic semiconductor layer. Consequently, the injected charge carrier (into the organic semiconductor layer) may recombine with the charge at the passive layer if the energy band of the charged organic semiconductor layer does not substantially change. Matching energy bands involves compromising between ease of charge injection and length of charge (data) retention time.

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In one embodiment, when matching energy bands, the fermi level of the passive layer is within about 0.3 eV of the valence band of the organic semiconductor layer. In another embodiment, the fermi level of the passive layer is within about 0.25 eV of the valence band of the organic semiconductor layer. In yet another embodiment, the fermi level of the passive layer is within about 0.2 eV of the valence band of the organic semiconductor layer. In still yet another embodiment, the fermi level of the passive layer is within about 0.15 eV of the valence band of the organic semiconductor layer.

The applied external field can reduce the energy barrier between passive layer and organic layer depending on the field direction. Therefore, enhanced charge injection in the forward direction field in programming operation and also enhanced charge recombination in reversed field in erase operation can be obtained.

The passive layer may in some instances act as a catalyst when forming the organic semiconductor layer, particularly when the organic semiconductor layer contains a conjugated organic polymer. In this connection, the polymer backbone of the conjugated organic polymer may initially form adjacent the passive layer, and grow or assemble away and substantially perpendicular to the passive layer surface. As a result, the polymer backbones of the conjugated organic polymers may be self aligned in a direction that traverses the two electrodes.

Examples of conductivity facilitating compounds that may make up the passive layer include one or more of copper sulfide (Cu₂S, CuS), copper rich copper sulfide (Cu₃S/Cu₂S, Cu₃S/CuS), copper oxide (CuO, Cu₂O), manganese oxide (MnO₂), titanium dioxide (TiO₂), indium oxide (I₃O₄), silver sulfide (Ag₂S, AgS), gold sulfide (Au₂S, AuS), iron oxide (Fe₃O₄), nickel arsenide (NiAs), cobalt arsenide (CoAs₂), and the like. The conductivity facilitating compounds do not dissociate into ions under the strength of the electric field. The passive layer may contain two or more sub-passive layers, each sub-layer containing the same, different, or multiple conductivity facilitating compounds.

The passive layer is grown using oxidation techniques, formed via gas phase reactions, or deposited between the electrodes. In some instances, to promote long

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charge retention times (in the organic semiconductor layer), the passive layer may be treated with a plasma after it is formed. The plasma treatment modifies the energy barrier of the passive layer.

In one embodiment, the passive layer containing the conductivity facilitating compound has a thickness of about 2 Å or more and about 0.1 µm or less. In another embodiment, the passive layer has a thickness of about 10 Å or more and about 0.01 µm or less. In yet another embodiment, the passive layer has a thickness of about 50 Å or more and about 0.005 µm or less.

In order to facilitate operation of the organic memory cells, the organic semiconductor layer is thicker than the passive layer. In one embodiment, the thickness of the organic semiconductor layer is from about 10 to about 500 times greater than the thickness of the passive layer. In another embodiment, the thickness of the organic semiconductor layer is from about 25 to about 250 times greater than the thickness of the passive layer.

The area size of the individual organic memory cells (as measured by the surface area of the two electrodes directly overlapping each other) can be small compared to conventional silicon based inorganic memory cells such as MOSFETs. In one embodiment, the area size of the organic memory cells of the present invention is about 0.0001 μ m² or more and about 4 μ m² or less. In another embodiment, the area size of the organic memory cells is about 0.001 μ m² or more and about 1 μ m² or less.

Operation of the organic memory devices/cells is facilitated using an external stimuli to achieve a switching effect. The external stimuli include an external electric field and/or light radiation. Under various conditions, the organic memory cell is either conductive (low impedance or "on" state) or non-conductive (high impedance or "off" state).

The organic memory cell may further have more than one conductive or low impedance state, such as a very highly conductive state (very low impedance state), a highly conductive state (low impedance state), a conductive state (medium level impedance state), and a non-conductive state (high impedance state) thereby enabling

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the storage of multiple bits of information in a single organic memory cell, such as 2 or more bits of information or 4 or more bits of information.

Switching the organic memory cell to the "on" state from the "off" state occurs when an external stimuli such as an applied electric field exceeds a threshold value. Switching the organic memory cell to the "off" state from the "on" state occurs when an external stimuli does not exceed a threshold value or does not exist. The threshold value varies depending upon a number of factor including the identity of the materials that constitute the organic memory cell and the passive layer, the thickness of the various layers, and the like.

Generally speaking, the presence of an external stimuli such as an applied electric field that exceeds a threshold value ("on" state) permits an applied voltage to write or erase information into/from the organic memory cell and the presence of an external stimuli such as an applied electric field that is less than a threshold value permits an applied voltage to read information from the organic memory cell; whereas the absence of the external stimuli that exceeds a threshold value ("off" state) prevents an applied voltage to write or erase information into/from the organic memory cell.

To write information into the organic memory cell, a voltage or pulse signal that exceeds the threshold is applied. To read information written into the organic memory cell, a voltage or electric field of any polarity is applied. Measuring the impedance determines whether the organic memory cell is in a low impedance state or a high impedance state (and thus whether it is "on" or "off"). To erase information written into the organic memory cell, a negative voltage or a polarity opposite the polarity of the writing signal that exceeds a threshold value is applied.

The organic memory devices described herein can be employed to form logic devices such as central processing units (CPUs); volatile memory devices such as DRAM devices, SRAM devices, and the like; input/output devices (I/O chips); and non-volatile memory devices such as EEPROMs, EPROMs, PROMs, and the like. The organic memory devices may be fabricated in planar orientation (two dimensional) or three dimensional orientation containing at least two planar arrays of the organic memory cells.

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Referring to Figure 7, a three dimensional microelectronic organic memory device 700 containing a plurality of organic memory cells in accordance with an aspect of the invention is shown. The three dimensional microelectronic organic memory device 700 contains a plurality of first electrodes 702, a plurality of second electrodes 704, and a plurality of memory cell layers 706. Between the respective first and second electrodes are the controllably conductive media (not shown). The plurality of first electrodes 702 and the plurality of second electrodes 704 are shown in substantially perpendicular orientation, although other orientations are possible. The three dimensional microelectronic organic memory device is capable of containing an extremely high number of memory cells thereby improving device density. Peripheral circuitry and devices are not shown for brevity.

The organic memory cells/devices are useful in any device requiring memory. For example, the organic memory devices are useful in computers, appliances, industrial equipment, hand-held devices, telecommunications equipment, medical equipment, research and development equipment, transportation vehicles, radar/satellite devices, and the like. Hand-held devices, and particularly hand-held electronic devices, achieve improvements in portability due to the small size and light weight of the organic memory devices. Examples of hand-held devices include cell phones and other two way communication devices, personal data assistants, palm pilots, pagers, notebook computers, remote controls, recorders (video and audio), radios, small televisions and web viewers, cameras, and the like.

Although the invention has been shown and described with respect to a certain embodiments, equivalent alterations and modifications will occur to others skilled in the art upon the reading and understanding of this specification and the annexed drawings. In regard to the various functions performed by the above described components (assemblies, devices, etc.), the terms used to describe such components are intended to correspond, unless otherwise indicated, to any component which performs the specified function of the described component (i.e., that is functionally equivalent), even though not structurally equivalent to the disclosed structure which performs the function in the herein illustrated exemplary embodiments of the

invention. In addition, while a particular feature of the invention may have been disclosed with respect to only one of several embodiments, such feature may be combined with one or more other features of the other embodiments as may be desired and advantageous for any given or particular application.